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Organic Chemistry

Robert Thornton Morrison Robert Neilson Boyd

New York University



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Prefa Ackn

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CHAP. 20

20.2 Nomenclature

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The names of acid derivatives are taken in simple ways from either the common name or the IUPAC name of the corresponding carboxylic acid. For example:

20.3 Physical properties

Ethyl ethanoate

The presence of the C=O group makes the acid derivatives polar compounds and appropriate (T-11 20) Acid chlorides and anhydrides (Table 20.1) and esters (Table 20.2, p. points that are about the boiling points that are about the same as those of aldehydes or ketones of comparable molecular weight (can be a second and anniverse of aldehydes or ketones of comparable molecular weight (can be a second and anniverse of aldehydes or ketones of comparable molecular weight (see Sec. 18.3). Amides (Table 20.1) have quite high boiling points because they are contained. boiling points because they are capable of strong intermolecular hydrogen bonding.

SEC. 20.4

The b the esters t the usual o Volati in the prep: irritating of acids.

> Acetyl cl Propiony. n-Butyryl (n-Valeryl (Stearoyl cl Benzoyl cl p-Nitrober chloride 3,5-Dinitro

Acetic anh Phthalic ar

chloride

 $20.4 - N_{\rm Hz}$

Before v

outline certa individual fa Each de correspondinacid by simpl conversion OI has certain cl $Th_{e\ deri}$ $^{carbonyl}\ grot$ undergone by But by its pre $\iota_{hese} \operatorname{compou}$ Here, to $f_{unctions}$: (a) acidity of hyd

(We shall 21.11-21.12 a